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Modern Chemistry 3 Chemical Equations
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Chemical Equations and Reactions

SECTION 8-1 SHORT ANSWER Answer

the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right.

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Chemical Equations And ...**

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CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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CHAPTER 8 REVIEW Chemical Equations
and Reactions SECTION 3 SHORT

ANSWER Answer the following questions
in the space provided. 1. List four metals
that will not replace hydrogen in an acid.
Choose from Cu, Ag, Au, Pt, Sb, Bi, and
Hg.

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Chemical Equations and Reactions

SHORT ANSWER Answer the following questions in the space provided. 1.

Match the symbol on the left with its appropriate description on the right. (aq)
(a) (b) (c) (d) (f) A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied to the reactants.

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CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form

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Equations and Reactions SECTION 3
SHORT ANSWER Answer the following questions in the space provided

1 List four metals that will not replace hydrogen in an acid Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg

2 Consider the metals iron and silver, both listed in Table 3 on page 286 of the text Which one Covalent Bonding

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